

Cerium	Am	nmonium Nitrate	Pricing >
Linear Formula	$Ce(NH_4)_2(NO_3)_6$		
Pubchem CID	16211559		
MDL Number	MFCD00151121		
EC No.	240-827-6		
IUPAC Name	azane; cerium(4+); nitric acid; tetranitrate		
Beilstein/Reaxys No.	N/A		
SMILES	[Cd].[Sb]		
Inchl Identifier	InChl=1S/Ce.2HNO3.4NO3.2H3N/c;6*2-1(3)4;;/h;2*(H,2,3,4);;;;2*1H3/q+4;;;4*-1;;		
Inchi Key	WIBGOERAEYJBOT-UHFFFAOYSA-N		
Signal Word		Danger	
Hazard Statements		H272-H315-H319-H335	
Hazard Codes		O, Xi	
Precautionary Statements		P221-P210v-P220-P305+P351+P338-P405-P501a	
Risk Codes		R8 R36/37/38	
Safety Statements		N/A	
RTECS Number		N/A	
Transport Information		UN1477 5.1/PG II	
WGK Germany		3	
GHS Pictograms		GHS03 Oxidizer GHS07 Exclamation	<u>Point</u>

Create Printable PDF

SAFETY DATA SHEET

Date Accessed: 04/20/2024 **Date Revised:** 01/15/2022

SECTION 1. IDENTIFICATION

Product Identifiers: All applicable American Elements product codes for CAS #16774-21-3

Relevant identified uses of the substance:

Scientific research and development

Supplier details:

American Elements 10884 Weyburn Ave. Los Angeles, CA 90024 Tel: +1 310-208-0551

Fax: +1 310-208-0351

Emergency telephone number: Domestic, North America +1 800-424-9300 International +1 703-527-3887

SECTION 2. HAZARDS IDENTIFICATION

Classification of the substance or mixture Classification according to Regulation (EC) No 1272/2008

GHS03 Flame over circle

Ox. Sol. 2 H272 May intensify fire; oxidizer.

GHS07

Skin Irrit. 2 H315 Causes skin irritation.

Eye Irrit. 2A H319 Causes serious eye irritation.

STOT SE 3 H335 May cause respiratory irritation.

Hazards not otherwise classified

No data available

Label elements

Labelling according to Regulation (EC) No 1272/2008 The substance is classified and labeled according to the CLP regulation.

Hazard pictograms





GHS03 GHS07

Signal word: Danger Hazard statements

H272 May intensify fire; oxidizer.

H315 Causes skin irritation.

H319 Causes serious eye irritation.

H335 May cause respiratory irritation.

Precautionary statements

P221 Take any precaution to avoid mixing with combustibles.

P210 Keep away from heat/sparks/open flames/hot surfaces. - No smoking.

P220 Keep/Store away from clothing/combustible materials.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P405 Store locked up.

P501 Dispose of contents/container in accordance with local/regional/national/international regulations. WHMIS classification

C - Oxidizing materials

D2B - Toxic material causing other toxic effects

Classification system

HMIS ratings (scale 0-4)

(Hazardous Materials Identification System)

HEALTH

FIRE

REACTIVITY

1

0

2

Health (acute effects) = 1

Flammability = 0

Physical Hazard = 2

Other hazards

Results of PBT and vPvB assessment

PBT: N/A vPvB: N/A

SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS

Substances

CAS No. / Substance Name:

16774-21-3 Cerium(IV) ammonium nitrate

Identification number(s): EC number: 240-827-6

SECTION 4. FIRST AID MEASURES

Description of first aid measures

If inhaled:

Supply patient with fresh air. If not breathing, provide artificial respiration. Keep patient warm.

Seek immediate medical advice.

In case of skin contact:

Immediately wash with soap and water; rinse thoroughly.

Seek immediate medical advice.

In case of eye contact:

Rinse opened eye for several minutes under running water. Consult a physician.

If swallowed:

Seek medical treatment.

Information for doctor

Most important symptoms and effects, both acute and delayed

No data available

Indication of any immediate medical attention and special treatment needed

No data available

SECTION 5. FIREFIGHTING MEASURES

Extinguishing media

Suitable extinguishing agents

Carbon dioxide, extinguishing powder or water spray. Fight larger fires with water spray or alcohol resistant foam.

For safety reasons unsuitable extinguishing agents Halocarbon extinguisher

Special hazards arising from the substance or mixture This substance is an oxidizer and its heat of reaction with reducing agents or combustibles may cause ignition.

If this product is involved in a fire, the following can be released:

Nitrogen oxides (NOx)

Metal oxide fume

Ammonia

Advice for firefighters

Protective equipment:

Wear self-contained respirator.

Wear fully protective impervious suit.

SECTION 6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Use personal protective equipment. Keep unprotected persons away.

Ensure adequate ventilation

Environmental precautions:

Do not allow material to be released to the environment without official permits.

Methods and materials for containment and cleanup:

Ensure adequate ventilation.

Prevention of secondary hazards:

Acts as an oxidizing agent on organic materials such as wood, paper and fats

Keep away from combustible material.

Reference to other sections

See Section 7 for information on safe handling

See Section 8 for information on personal protection equipment.

See Section 13 for disposal information.

SECTION 7. HANDLING AND STORAGE

Handling

Precautions for safe handling

Keep container tightly sealed.

Store in cool, dry place in tightly closed containers.

Ensure good ventilation at the workplace.

Information about protection against explosions and fires:

Substance/product can reduce the ignition temperature of flammable substances.

This substance is an oxidizer and its heat of reaction with reducing agents or combustibles may cause ignition.

Conditions for safe storage, including any incompatibilities

Requirements to be met by storerooms and receptacles:

No special requirements.

Information about storage in one common storage facility:

Store away from flammable substances.

Store away from reducing agents.

Do not store with organic materials.

Do not store together with acids.

Store away from strong bases.

Store away from metal powders.

Further information about storage conditions:

Keep container tightly sealed.

Store in cool, dry conditions in well-sealed containers.

Specific end use(s)

No data available

SECTION 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Additional information about design of technical systems:

Properly operating chemical fume hood designed for hazardous chemicals and having an average face velocity of at least 100 feet per minute.

Control parameters

Components with limit values that require monitoring at the workplace:

None.

Additional information: No data

Exposure controls

Personal protective equipment

Follow typical protective and hygienic practices for handling chemicals.

Keep away from foodstuffs, beverages and feed.

Remove all soiled and contaminated clothing immediately.

Wash hands before breaks and at the end of work.

Avoid contact with the eyes and skin.

Maintain an ergonomically appropriate working environment.

Breathing equipment:

Use suitable respirator when high concentrations are present.

Recommended filter device for short term use: Use a respirator with type N95 (USA) or PE (EN 143) cartridges as a backup to engineering controls. Risk assessment should be performed to determine if air-

purifying respirators are appropriate. Only use equipment tested and approved under appropriate government standards.

Protection of hands:

Impervious gloves

Inspect gloves prior to use.

Suitability of gloves should be determined both by material and quality, the latter of which may vary by manufacturer.

Material of gloves
Nitrile rubber, NBR

Penetration time of glove material (in minutes)

480

Glove thickness

0.11 mm

Eye protection: Safety glasses

Body protection: Protective work clothing.

SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties Appearance:

Form: Various forms (powder/flake/crystalline/beads,

etc.)

Color: Yellow to orange

Odor: Pungent

Odor threshold: No data available.

pH: N/A

Melting point/Melting range: No data available Boiling point/Boiling range: No data available Sublimation temperature / start: No data available

Flammability (solid, gas)

Contact with combustible material may cause fire.

Ignition temperature: No data available

Decomposition temperature: No data available

Autoignition: No data available.

Danger of explosion: No data available.

Explosion limits:

Lower: No data available Upper: No data available Vapor pressure: N/A Density: No data available

Relative density: No data available.

Vapor density: N/A Evaporation rate: N/A Solubility in / Miscibility with Water at 20 °C (68 °F): 1410 g/l

Partition coefficient (n-octanol/water): No data

available.
Viscosity:
Dynamic: N/A
Kinematic: N/A
Other information
No data available

SECTION 10. STABILITY AND REACTIVITY

Reactivity

May intensify fire; oxidizer.

Chemical stability

Stable under recommended storage conditions.

Thermal decomposition / conditions to be avoided:

Decomposition will not occur if used and stored

according to specifications.

Possibility of hazardous reactions

Reacts with reducing agents

Reacts with flammable substances

Conditions to avoid

No data available

Incompatible materials:

Acids

Flammable substances

Reducing agents

Bases

Organic materials

Metal powders

Hazardous decomposition products:

Nitrogen oxides

Metal oxide fume

Ammonia

SECTION 11. TOXICOLOGICAL INFORMATION

Information on toxicological effects Acute toxicity: No effects known.

LD/LC50 values that are relevant for classification: No

data

Skin irritation or corrosion: Causes skin irritation. Eye irritation or corrosion: Causes serious eye

irritation.

Sensitization: No sensitizing effects known. Germ cell mutagenicity: No effects known.

Carcinogenicity:

No classification data on carcinogenic properties of this material is available from the EPA, IARC, NTP, OSHA or ACGIH.

Reproductive toxicity: No effects known.

Specific target organ system toxicity - repeated

exposure: No effects known.

Specific target organ system toxicity - single exposure: May cause respiratory irritation. Aspiration hazard: No effects known.

Subacute to chronic toxicity: No effects known.

Additional toxicological information:

To the best of our knowledge the acute and chronic

toxicity of this substance is not fully known.

Carcinogenic categories

OSHA-Ca (Occupational Safety & Health

Administration)

Substance is not listed.

SECTION 12. ECOLOGICAL INFORMATION

Toxicity

Aquatic toxicity:

No data available

Persistence and degradability

No data available

Bioaccumulative potential

No data available

Mobility in soil

No data available

Additional ecological information:

Do not allow material to be released to the

environment without official permits.

Avoid transfer into the environment.

Results of PBT and vPvB assessment

PBT: N/A vPvB: N/A

Other adverse effects No data available

SECTION 13. DISPOSAL CONSIDERATIONS

Waste treatment methods

Recommendation

Consult official regulations to ensure proper disposal.

Uncleaned packagings:

Recommendation:

Disposal must be made according to official regulations.

SECTION 14. EXPOSURE CONTROLS/PERSONAL PROTECTION

UN-Number

DOT, IMDG, IATA

UN1477

UN proper shipping name

DOT

Nitrates, inorganic, n.o.s. (Cerium(IV) ammonium nitrate)

IMDG, IATA

NITRATES, INORGANIC, N.O.S. (Cerium(IV)

ammonium nitrate)

Transport hazard class(es)

DOT

Class

5.1 Oxidising substances.

Label

5.1

Class

5.1 (O2) Oxidizing substances

Label

5.1

IMDG, IATA

Class

5.1 Oxidising substances.

Label

5.1

Packing group

DOT, IMDG, IATA

П

Environmental hazards:

N/A

Special precautions for user

Warning: Oxidizing substances

EMS Number:

F-A,S-Q

Transport in bulk according to Annex II of

MARPOL73/78 and the IBC Code

N/A

Transport/Additional information:

DOT

Marine Pollutant (DOT):

No

UN "Model Regulation":

UN1477, Nitrates, inorganic, n.o.s. (Cerium(IV)

ammonium nitrate), 5.1, II

SECTION 15. REGULATORY INFORMATION

Safety, health and environmental regulations/legislation specific for the substance or

mixture

National regulations

All components of this product are listed in the U.S. Environmental Protection Agency Toxic Substances

Control Act Chemical substance Inventory.

All components of this product are listed on the Canadian Domestic Substances List (DSL).

SARA Section 313 (specific toxic chemical listings)

Substance is not listed.

California Proposition 65

Prop 65 - Chemicals known to cause cancer

Substance is not listed.

Prop 65 - Developmental toxicity

Substance is not listed.

Prop 65 - Developmental toxicity, female

Substance is not listed.

Prop 65 - Developmental toxicity, male

Substance is not listed.

Information about limitation of use:

For use only by technically qualified individuals.

Other regulations, limitations and prohibitive regulations

Substance of Very High Concern (SVHC) according to the REACH Regulations (EC) No. 1907/2006.

Substance is not listed.

The conditions of restrictions according to Article 67 and Annex XVII of the Regulation (EC) No 1907/2006 (REACH) for the manufacturing, placing on the market and use must be observed.

Substance is not listed.

Annex XIV of the REACH Regulations (requiring

Authorisation for use)

Substance is not listed.

Chemical safety assessment:

A Chemical Safety Assessment has not been carried out.

16. OTHER INFORMATION

Safety Data Sheet according to Regulation (EC) No. 1907/2006 (REACH). The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. American Elements shall not be held liable for any damage resulting from handling or from contact with the above product. See reverse side of invoice or packing slip for additional terms and conditions of sale. COPYRIGHT 1997-2022 AMERICAN ELEMENTS. LICENSED GRANTED TO

Reseach

- Generation and cross-coupling of benzyl and phthalimide-N-oxyl radicals in a cerium(IV) ammonium nitrate/N-hydroxyphthalimide/ArCH2R system. Alexander O. Terent'ev, Igor B. Krylov, Mikhail Y. Sharipov, Zoya M. Kazanskaya, Gennady I. Nikishin. Tetrahedron, Volume 68, Issue 50, 16 December 2012, Pages 10263-10271.
- Solvent free-synthesis of highly functionalized 4H-chromene-3carboxamide derivatives using cerium ammonium nitrate and their antioxidant, antibacterial and solvatochromism studies. Subbareddy V. Chitreddy, Sumathi Shanmugam. Journal of Molecular Liquids, Volume 243, October 2017, Pages 494-502.
- Cerium ammonium nitrate-mediated the oxidative dimerization of palkenylphenols: a new synthesis of substituted (±)-transdihydrobenzofurans. Po-Yuan Chen, Yi-Hua Wu, Mon-Huei Hsu, Tzu-Pin Wang, Eng-Chi Wang. Tetrahedron, Volume 69, Issue 2, 14 January 2013, Pages 653-657.
- Cerium ammonium nitrate: an efficient catalyst for carbon—carbon bond formation from ferrocenyl alcohol substrate. Xiaoping Xu, Ran Jiang, Xiaoguang Zhou, Yu Liu, Yong Zhang. Tetrahedron, Volume 65, Issue 4, 24 January 2009, Pages 877-882.
- Convenient synthesis of substituted pyrroles via a cerium (IV) ammonium nitrate (CAN)-catalyzed Paal–Knorr reaction. Ahmed Kamal, Shaikh Faazil, M. Shaheer Malik, Moku Balakrishna, Abdullah Alarifi. Arabian Journal of Chemistry, Volume 9, Issue 4, July 2016, Pages 542-549.
- Cerium (IV) ammonium nitrate-mediated reactions: Simple route to benzimidazole derivatives. Kamal Usef Sadek, Fawzia Al-Qalaf, Ramadan Ahmed Mekheimer, Mohamed Hilmy Elnagdi. Arabian Journal of Chemistry, Volume 5, Issue 1, January 2012, Pages 63-66.
- Ultrasound-assisted facile one-pot sequential synthesis of novel sulfonamide-isoxazoles using cerium (IV) ammonium nitrate (CAN) as an efficient oxidant in aqueous medium. Soukaina Alaoui, Mohsine Driowya, Luc Demange, Rachid Benhida, Khalid Bougrin. Ultrasonics Sonochemistry, Volume 40, Part A, January 2018, Pages 289-297.
- Cerium(IV) ammonium nitrate: Reagent for the versatile oxidative functionalization of styrenes using N-hydroxyphthalimide. Igor B. Krylov, Stanislav A. Paveliev, Olesya K. Matveeva, Alexander O. Terent'ev. Tetrahedron, Volume 75, Issue 17, 26 April 2019, Pages 2529-2537.
- Cerium ammonium nitrate (CAN) for mild and efficient reagent to remove hydroxyethyl units from 2-hydroxyethyl ethers and 2hydroxyethyl amines. Hiromichi Fujioka, Hideki Hirose, Yusuke Ohba, Kenichi Murai, Yasuyuki Kita. Tetrahedron, Volume 63, Issue 3, 15 January 2007, Pages 625-637.
- Nucleophilic substitution of ferrocenyl alcohols by cerium ammonium

- nitrate: C–N, C–S, and C–O bonds formation. Ran Jiang, Ying Zhang, Ye-Chen Shen, Xu Zhu, Shun-Jun Ji. Tetrahedron, Volume 66, Issue 23, 5 June 2010, Pages 4073-4078.
- Cerium ammonium nitrate-mediated the oxidative dimerization of palkenylphenols: a new synthesis of substituted (±)-transdihydrobenzofurans. Po-Yuan Chen, Yi-Hua Wu, Mon-Huei Hsu, Tzu-Pin Wang, Eng-Chi Wang. Tetrahedron, Volume 69, Issue 2, 14 January 2013, Pages 653-657.